

MATERIAL SAFETY DATA SHEET FERRIC CHLORIDE LIQUID 40-42%

1. PRODUCT IDENTIFICATION

Common Name : Ferric Chloride
Chemical Family : Inorganic Salt
Chemical Name of major constituent : Ferric Chloride
Formula : FeCl
Molecular Weight : 162.25
UN Number : 2582
Exposure Limits :
OSHA TLV : Img/m³ (8 hours) TWA (as soluble iron salts)
ACGIH TLV : Img/m³ (8 hours) TWA (as soluble iron salts)
NIOSH recommended : Img/m³ (8 hours) TWA (as soluble iron salts)
Uses:- Potable and waste water treatment, odor removal, adhesive for dye, textile impression pigment, ink and photoengraving.

2. PHYSICAL AND CHEMICAL PROPERTIES

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Physical State Liquid
Appearance Reddish Brown
Melting/Freezing Point50° C (-58°F)
Boiling Point 105-110°C (220°F – 230°F)
Odor Slight Pungent Odor
Vapour Pressure N.A.
Specific Gravity (H ₂ O) 1.40 min.
Flash Point Non-flammable
Flammability Limit N.A.
Solubility in water Infinite
pH: < 1
Chemical Property:
Stability: Stable Hazardous
Decomposition Produces HCl Hazardous
Polymerization Does not occur Incompatible
Materials Metals and alloys
Condition to avoid Extensive heat, contact with strong

mineral acids and base or alkalis.

3. COMPOSITION

CHEMICAL NAME	CAS NUMBER	WEIGHT %
FeCl ₃	7705-08-0	42% min.
FeCl ₂	7758-94-3	< 0.5 %
HCl	7647-01-0	< 0.5 %

4. EXPOSURE CONTROL/PERSONAL PROTECTION

PERSONAL PROTECTION:

Ventilation : There should be enough local ventilation to keep the TLV

below ACGIH limits.

Maximum use limit.....: 500 mg/L

Respirator : Use an approved respirator with acid mist cartridges, if

necessary.

Eye Protection: Wear chemical goggles or face shields.

Gloves: Use neoprene or equivalent. Never use leather.

Clothing: Protective clothing if necessary, should be neoprene of

equivalent.

Other

When cleaning, maintenance or any other situations when airborne contaminants and/or dust could be generated, use protective equipment to protect against ingestion or inhalation.

5. FIRST AID MEASURES

Skin contact: Wash with soap and water. Remove any contaminated clothing

and wash before reuse. If irritation or illness develops,

seek medical attention.

Eye contact: Flush eyes thoroughly with water for at least 15

minutes, taking care to keep eye lids opened to be sure that the rinsing is complete. Get medical attention.

Inhalation: Move to fresh air. Give artificial respiration if breathing has

stopped. If breathing is difficult, give oxygen. Get medical

attention

 $Ingestion \hspace{0.5cm}:\hspace{0.5cm} If \hspace{0.5cm} conscious, \hspace{0.5cm} drink \hspace{0.5cm} water \hspace{0.5cm} or \hspace{0.5cm} milk \hspace{0.5cm} of \hspace{0.5cm} magnesia. \hspace{0.5cm} \underline{DO \hspace{0.5cm} NOT \hspace{0.5cm} induce}$

vomiting and do not give bicarbonate to neutralize. Get medical

attention.

6. FIRE FIGHTING MEASURES

Flash Point and MethodNot Applicable.

GENERAL HAZARD: This product is not flammable, but exposing to high

temperature may produce Hydrogen Chloride

vapours.

Extinguishing Media $\,\,$: Use water spray, fog, foam, dry chemical, CO $_2$ or

other agents as appropriate for surrounding fire.

Special Fire Fighting
Procedure

Wear self contained breathing apparatus and full
protective clothing as appropriate for surrounding

fire. Cool exterior of storage tanks.

7. ACCIDENTAL RELEASE MEASURES

Wear appropriate protective gear (See Section 4).

Contain and eliminate the release in order to prevent contamination of land or water way; neutralize with lime, lime stone or soda ash. This will generate CO₂, so additional ventilation may be necessary. Collect the residue for proper disposal or flush with water in accordance with applicable regulations to waste treatment system.

8. HAZARD IDENTIFICATION:

Ecological Information : Not available.

Principal Risk : WHMIS - Class E. Irritating to skin, eyes and

mucous membranes.

Potential Effects on Health: Acute and Chronic.

 $\label{lem:carcinogens} Carcinogenicity \quad : Does \ not \ contain \ any \ carcinogens \ or \ potential \ carcinogens.$

9. DISPOSAL CONSIDERATIONS

This material exhibits the RCRA characteristic of corrosivity. Ferric Chloride for disposal should be flushed out with plenty of water to dilute the solution

10. TRANSPORT INFORMATION

Shipping Name	: Ferric III Chloride Solution
Hazardous Class	: 8 (9.2)
DOT Number	: UN 2582
IMDG Code Page No	: 8173
Packing Group	: III

11. HANDLING AND STORAGE

Wear appropriate protective equipment (See Section 4). Store away from heat, strong alkalies and alkali metals. Keep containers closed and dry. Protect container from physical damage. Use handling equipment (pumps, hoses, etc.) compatible with product, ie., polyethylene, polypropylene, PVC, Teflon, rubber, FRP and titanium. Avoid contact with bare metals other than titanium. Wash thoroughly after handling. Emptied container may retain vapour and product residue.

DubaiChemical.com provides the foregoing information in good faith and makes no representations as to its comprehensiveness or accuracy. This document is intended only as a guide to the appropriate precautionary handling of the material by a properly trained person using the product. Individuals receiving the information must exercise their independent judgment in determining its appropriateness for a particular purpose.